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Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

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To cite this Article Reichardt, Bastian and Meier, Chris(2007) 'A New and Short Convergent Synthetic Strategy to Carbocyclic Nucleosides', *Nucleosides, Nucleotides and Nucleic Acids*, 26: 8, 935 — 937

To link to this Article: DOI: 10.1080/15257770701507937

URL: <http://dx.doi.org/10.1080/15257770701507937>

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A NEW AND SHORT CONVERGENT SYNTHETIC STRATEGY TO CARBOCYCLIC NUCLEOSIDES

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□ *An efficient synthesis for racemic cyclopent-3-en-1-yl nucleoside analogues has been developed starting from cyclopentadiene. The key step is the regioselective hydroboration of a mixture of intermediate alkylated cyclopentadienes to give one cyclopentenol.*

Keywords Carbocyclic nucleosides; cyclopentenyl nucleosides; Mitsunobu coupling; purines; pyrimidines

INTRODUCTION

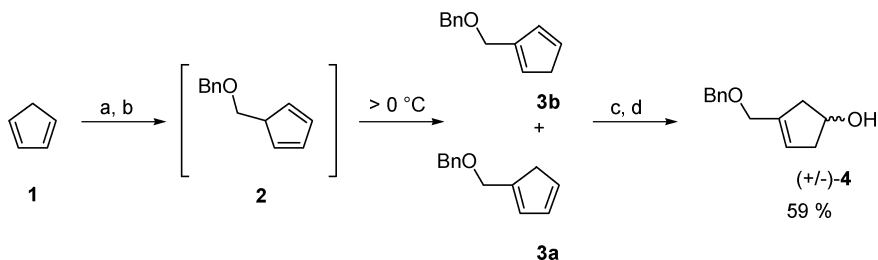
Carbocyclic nucleosides are compounds in which the furan ring has been replaced by a carbocyclic system. These nucleoside analogues have attracted considerable interest due to their important biological activity.^[1] Various structural modifications were introduced in both the heterocyclic base and the sugar moiety.^[2] Such compounds are stable to hydrolysis by phosphorylases that cleave the glycosidic bond in conventional nucleosides.^[3] The bioactivity of the naturally occurring carbocyclic nucleosides (-)-aristeromycin and (-)-neplanocin A led to an interest in this class of compounds, resulting in the generation of a number of antiviral carbocyclic nucleosides. In the past, carbocyclic nucleoside analogues like Abacavir^[4] (Ziagen) and Entecavir^[5] (Baraclude) showed encouraging in vitro and in vivo activities. Recently, Abacavir was approved as a HIV-drug for clinical application. Moreover, Entecavir was approved by FDA in 2005 for the treatment of chronic HBV infections.

RESULTS AND DISCUSSION

Recently, we published a new procedure for the preparation of racemic cyclopent-3-en-1-yl nucleoside analogues.^[7] Cyclopentenol **4** was prepared

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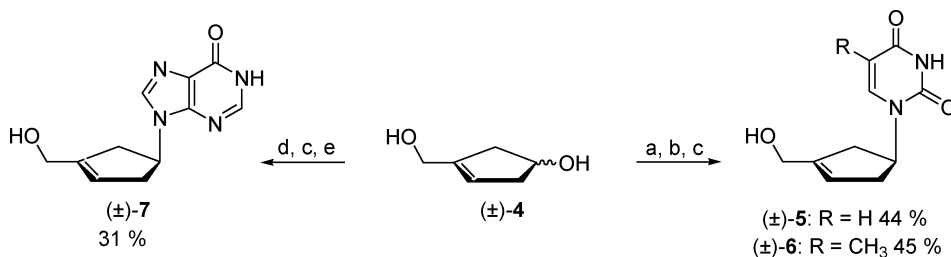
for a convergent approach to carbocyclic nucleosides. This cyclopentenol **4** was developed from cyclopentadiene **1** that was deprotonated and alkylated with benzyloxymethyl chloride to give the diene **2**. This material undergoes isomerization into *two* thermodynamically more stable cyclopentadienes **3a,b**. These two dienes were hydroborated with 9-BBN as a sterically demanding hydroboration reagent to (\pm)-3-[(benzyloxy)methyl]cyclopent-3-en-1-ol **4** in 42% overall yield. Next, the diene mixture was treated with 0.3 equiv. of BH_3 in THF. Surprisingly, also in this case cyclopentenol **4** was isolated as the sole product in 59% overall yield after alkaline work-up. (Scheme 1).



SCHEME 1 a) NaH , 0°C , 0.5 hours, THF; b) benzyloxymethyl chloride, -50°C , 2 hours, rt, 12 hours, THF; c) $\text{BH}_3 \cdot \text{THF}$, 24 hours, THF; d) $\text{NaOH}/\text{H}_2\text{O}_2$ 12 hours, THF.

To prove the suitability of cyclopentenol to act as starting material for the synthesis of carbocyclic nucleosides, pyrimidine heterocycles were introduced using the modified Mitsunobu reaction conditions as reported previously.^[8,11] Thus, *N*3-benzoylthymine and *N*3-benzoyluracil were condensed successfully with cyclopentenol **4**. Alkaline treatment of the crude reaction product led to *O*-benzylcyclopentenyl nucleosides. In the case of 3-benzoylthymine the *N*1/*O*²-regioselectivity was found to be 5:1 while a 9:1-regioselectivity was found for 3-benzoyluracil. Debenzylation of the nucleosides proceeded in 76–82% yield using BCl_3 at -78°C to give cyclopentenyl nucleoside analogues, respectively.^[12] Mitsunobu coupling of precursors (\pm)-**4** with 6-chloropurine led to the formation of the protected purine nucleosides in 54% yield, respectively.^[7] The *N*9/*N*7-regioselectivity was 9:1 and the two isomers were separated by column chromatography. Finally, removal of the *O*-benzyl group of the nucleoside proceeded in 82% yield using BCl_3 at -78°C to give cyclopentenyl nucleoside analogues **5,6**. In addition, the purine analogue was converted to the inosine nucleoside analogue (\pm)-**7** by treatment of the chloropurine derivative with sodium methanolate and 2-mercaptoethanol in 68% yield (Scheme 2).

In summary, we have elaborated an interesting new pathway towards the synthesis of carbocyclic pyrimidine and hypoxanthine nucleoside analogues. The method starts from cheap cyclopentadiene and is a powerful addition



SCHEME 2 a) DIAD, PPh_3 , $\text{N}^3\text{-}^{\text{Bz}}\text{Ura}/\text{N}^3\text{-}^{\text{Bz}}\text{Thy}$, -40°C , 2 hours, CH_3CN ; b) $\text{NaOH}/\text{CH}_3\text{OH}$ 1%, 12 hours; c) BCl_3 5 hours, -78°C , CH_2Cl_2 ; d) DIAD, PPh_3 , 6-chloropurine, 24 hours, THF; e) 2-mercaptoethanol, NaOMe , MeOH , 60°C , 4 hours.

to our previously reported methodology. The biological evaluation of the compounds is under investigation.

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